

Periodic DFT calculations for the solid state: Phonons and Vibrations in molecular crystals

Mark JOHNSON
Scientific Computing
Institut Laue Langevin
(<http://www.ill.fr/Computing/Club.html>)
Grenoble, France

Plan Of Talk

- I. Introduction to spectroscopy: IR, neutrons and X-rays
- II. *Hexamethyl benzene* – the importance of k-point sampling
- III. *Oligothiophenes* – from INS to IR
- IV. *Benzoic acid* – hydrogen bonded dimers, INS to IXS
- V. *Urea* - multiphonons for small molecules
- VI. *Urea phosphoric acid* – short hydrogen bonds
- VI. Conclusion & perspectives (collagen, DNA ...)

INS vs. IXS vs. IR

	INS	IXS	IR
Scattering by	Nuclei	Electrons	Electrons
Spectral intensities from eigenvectors and	Scattering lengths	Number of electrons and form factor	Variation of dipole moment via charge tensors
Selection rules	No	No	Yes
Phonon dispersion	Yes (K from $k_i, k_f, 2\theta$)	Yes (K from $2\theta, k_i=k_f$)	No ($K=0$)
Effect of isotopic substitution	Mass and scattering length	Mass only	Mass only
Resolution	Varies with energy transfer (> ~2%)	Constant (~meV)	Constant (~cm ⁻¹)
Counting times	Hours	Days	Seconds/minutes

INS vs. IXS vs. IR

3.46 Create/Neutron Scattering/Coherent Form Factor

3.49 Create/Neutron Scattering/Incoherent on Mono.

The
exp

PHONON calculates the one-phonon double differential incoherent neutron scattering cross section as a sum of contributions from all atoms of the unit cell

wh
as

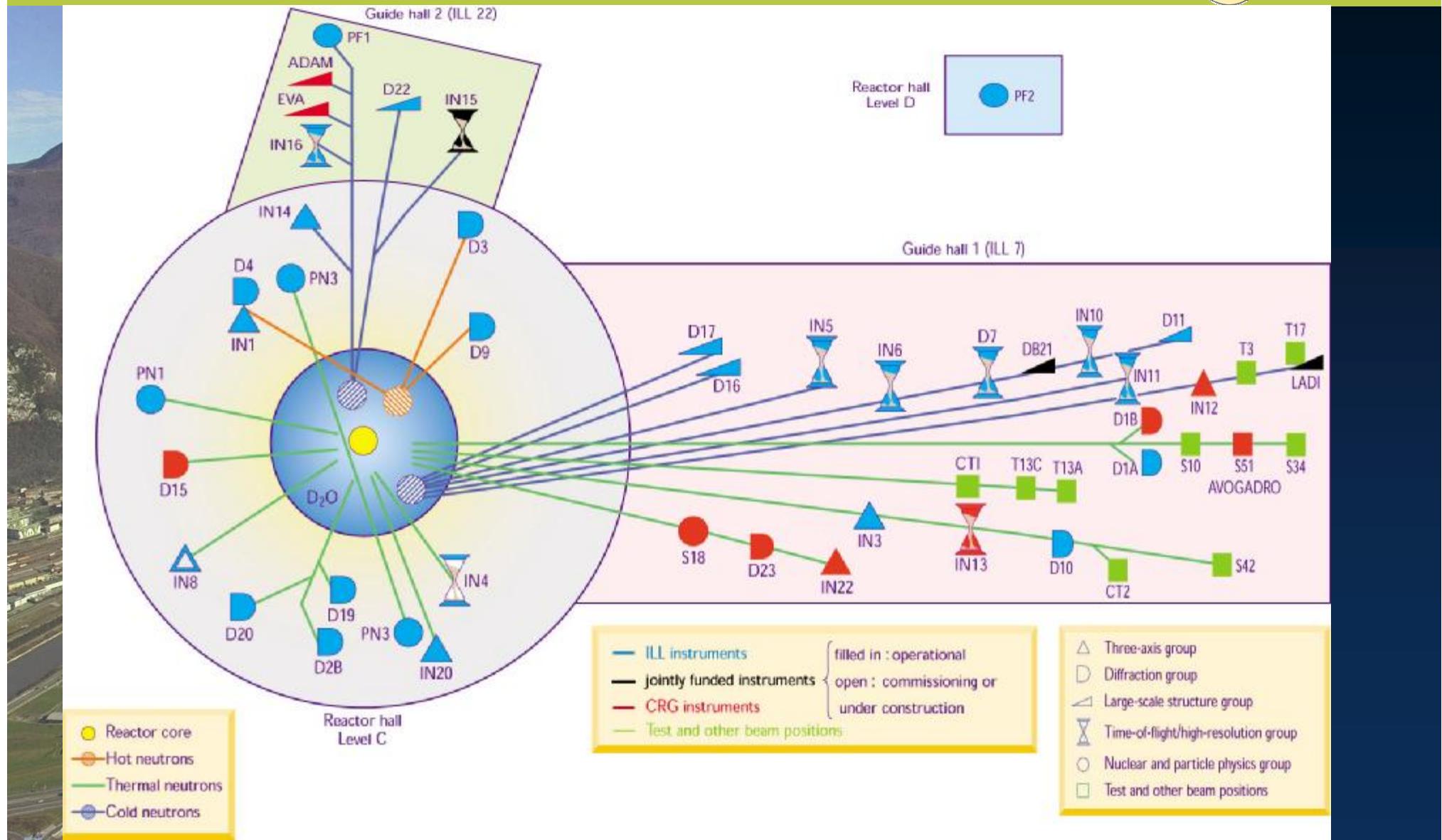
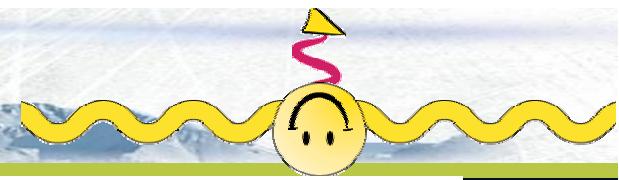
$$\frac{d^2\sigma_{inc}}{d\Omega dE} = \sum_{\mu} \frac{d^2\sigma_{inc}}{d\Omega dE}(\mu) \quad (88)$$

The contribution from atom μ is

$$\begin{aligned} \frac{d^2\sigma_{inc}}{d\Omega dE}(\mu) &= \sigma_{inc}(\mu) \frac{K}{K_o} \sum_{\mathbf{k}, j} \frac{\hbar}{2M_{\mu}\omega(\mathbf{k}, j)} |\boldsymbol{\kappa} \cdot \mathbf{e}(\mathbf{k}, j; \mu)|^2 \\ &\times (n(\omega(\mathbf{k}, j)) + 1) \exp[-2W_{\mu}(\kappa)] \delta(\epsilon - \omega(\mathbf{k}, j)) \end{aligned} \quad (89)$$

$$\text{IR: } abs_coeff(j) \sim \sum_g \left| \sum_{a,p} \frac{Z_{ga}^*(p)}{\sqrt{m_p}} e_a(p, j) \right|^2$$

NEUTRON SCATTERING AT ILL



2 ways to calculate a vibrational spectrum

Eigenvalue method

Matrix of force constants → dynamical matrix

DIAGONALISE → Eigenvalues (ω^2) & eigenvectors (e)

-> Vibrational density of states for all K -> INS, IXS, IR spectra

✓ Easy to visualise/understand modes, easy to calculate spectra for isotopomers

X But harmonic approximation, structure specific

Fourier transform method

Molecular dynamics simulation → velocity auto-correlation function or $I(K,t)$

FOURIER TRANSFORM → vDOS or $S(K,\omega)$ -> neutron spectrum

X More difficult to visualise modes, complete recalculation for isotopomers

✓ NO harmonic approximation

The computer model - structure and energy

Structure:

An ensemble of 100's atoms, (2000\AA^3)

Was single molecule or cluster (5-10 molecules),

Now unit cell (crystalline lattice) for $K=0$, supercell for $K \neq 0$ (for phonons)

Potential energy calculation:

VASP; *was* PW91 with USPP, *now* PBE with PAW

(key FLAGS; Prec=Accurate, Lreal=false, Ediff=1e-6, Addgrid=true,
+/- displacements of $\sim 0.05\text{\AA}$)

✓ convergence; plane wave energy cut-off & k-point spacing ($< 0.1\text{\AA}^{-1}$)

X BUT DFT underestimates (no!) dispersive interactions & effect of functional

Crystalline oligo-thiophenes (conducting polymers)

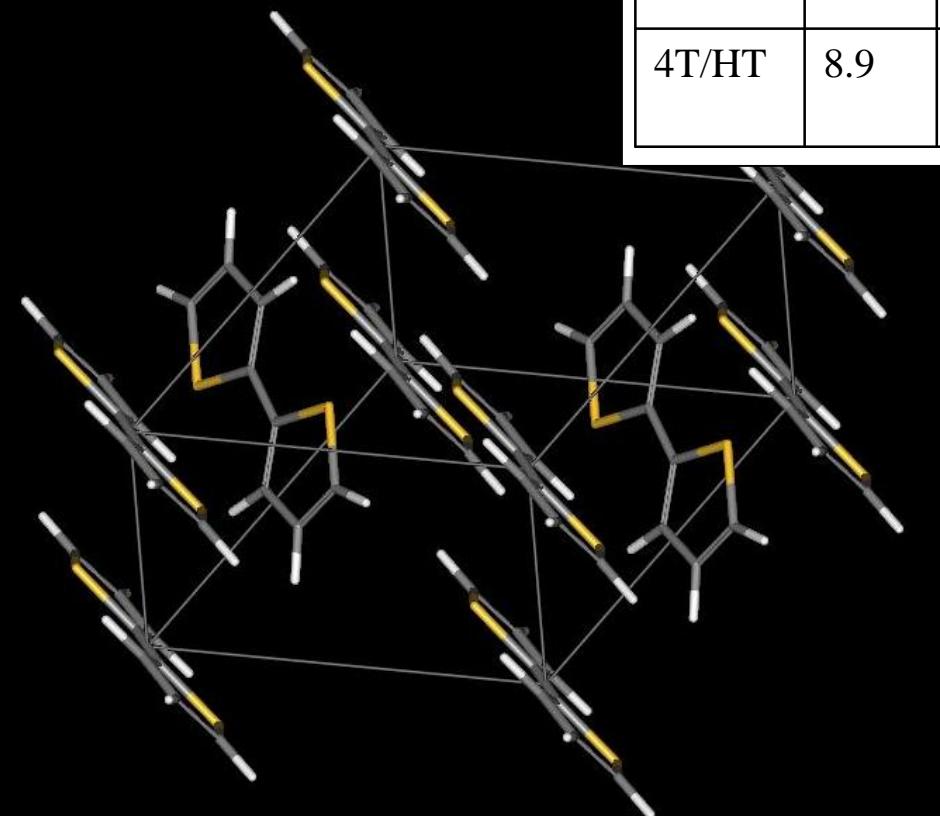
-From INS to IR

(Patrick Hermet, Martijn Marsman - VASP)

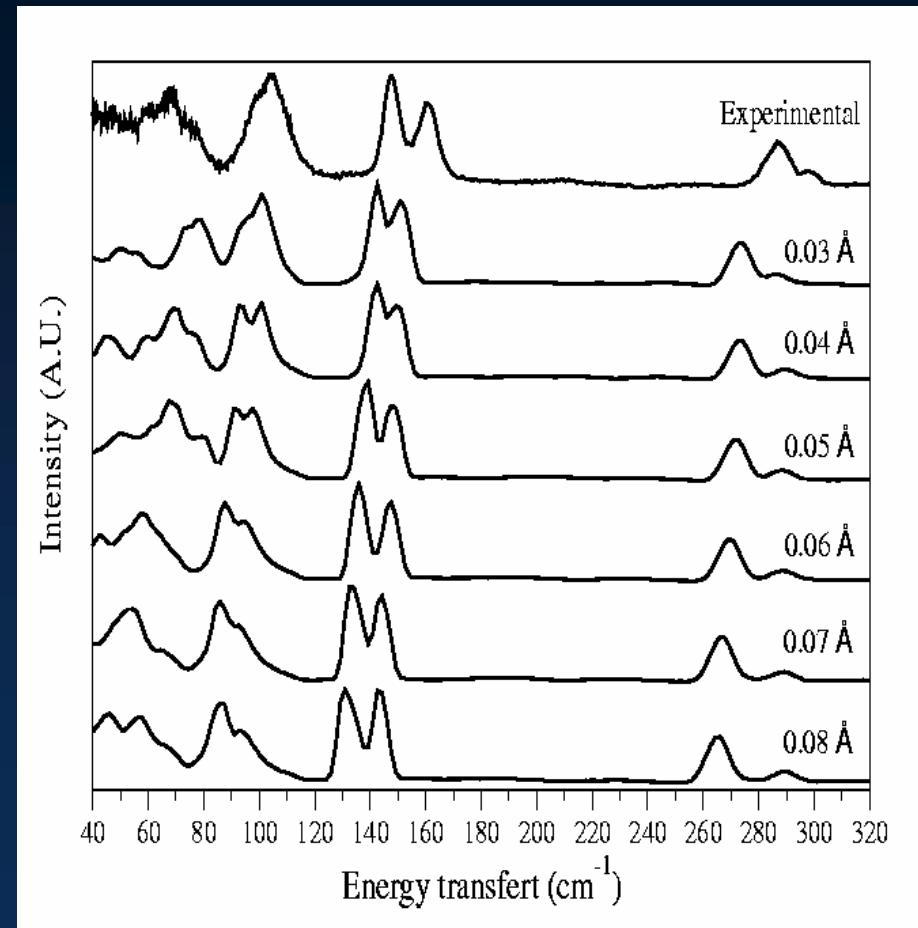
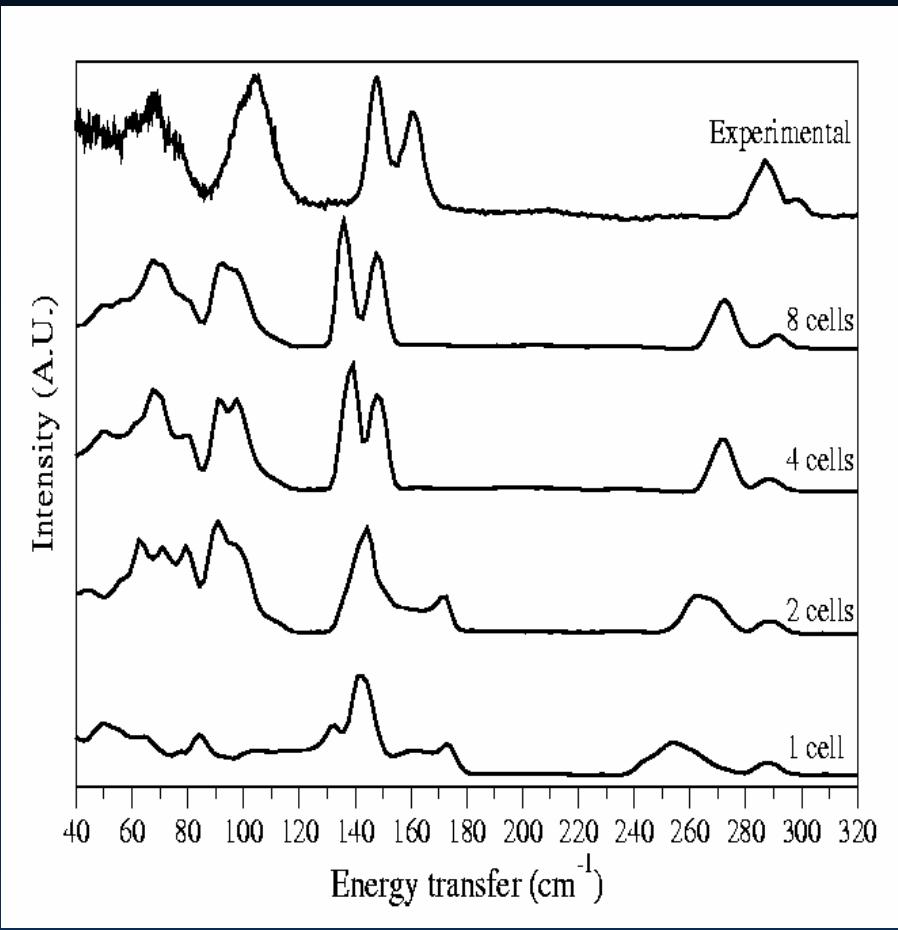
J.Phys:Cond.Matt 2004, J.Phys.Chem 2004

Thiophene crystal structures (monoclinic-P21/c)

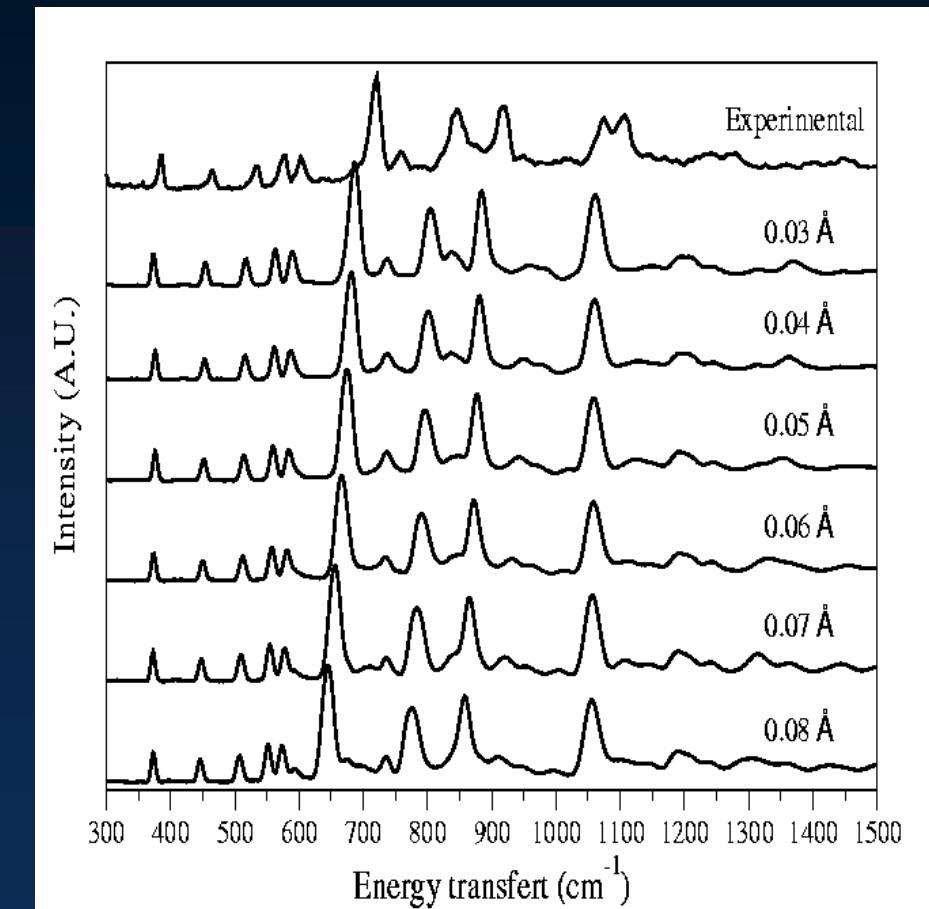
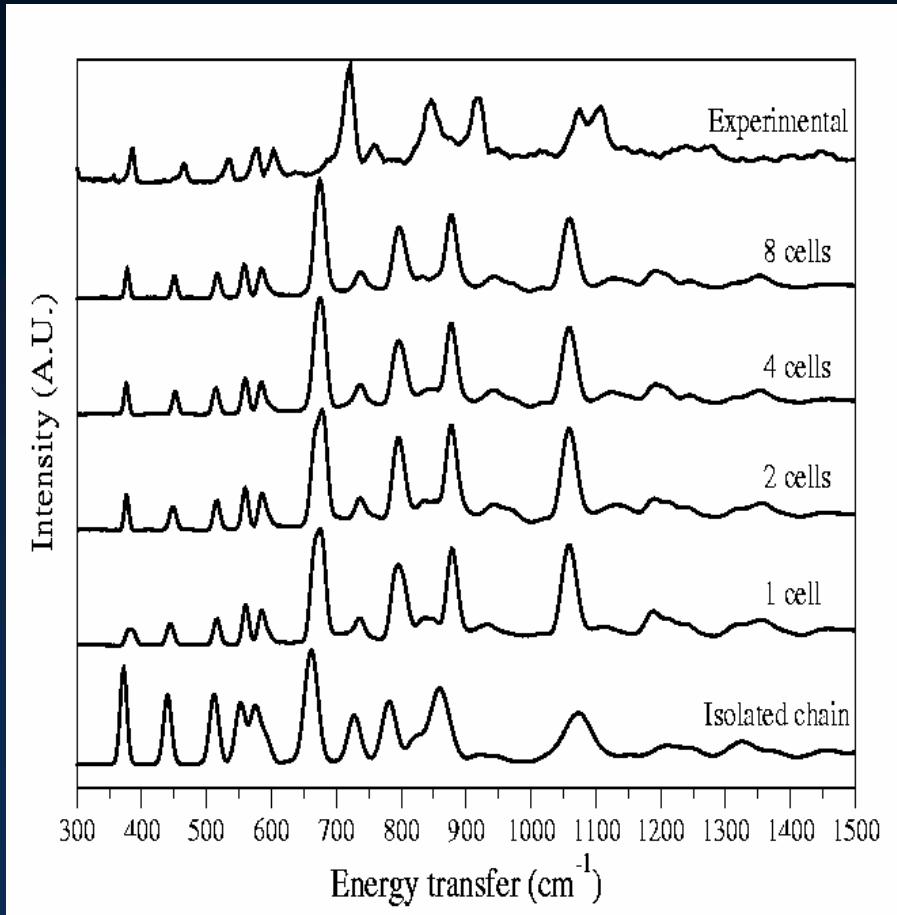
	a [Å]	b [Å]	c [Å]	β	Atoms/cell	Biggest supercell	Typical k-pts
2T	8.3	6.0	9.6	110	32 (2mol/cell)	2,2,2	2,2,2
4T/LT	6.1	7.9	30.5	92	120 (4mol/cell)	2,1,1	2,4,1
4T/HT	8.9	5.8	14.3	97	60 (2mol/cell)	1,2,1	2,2,2



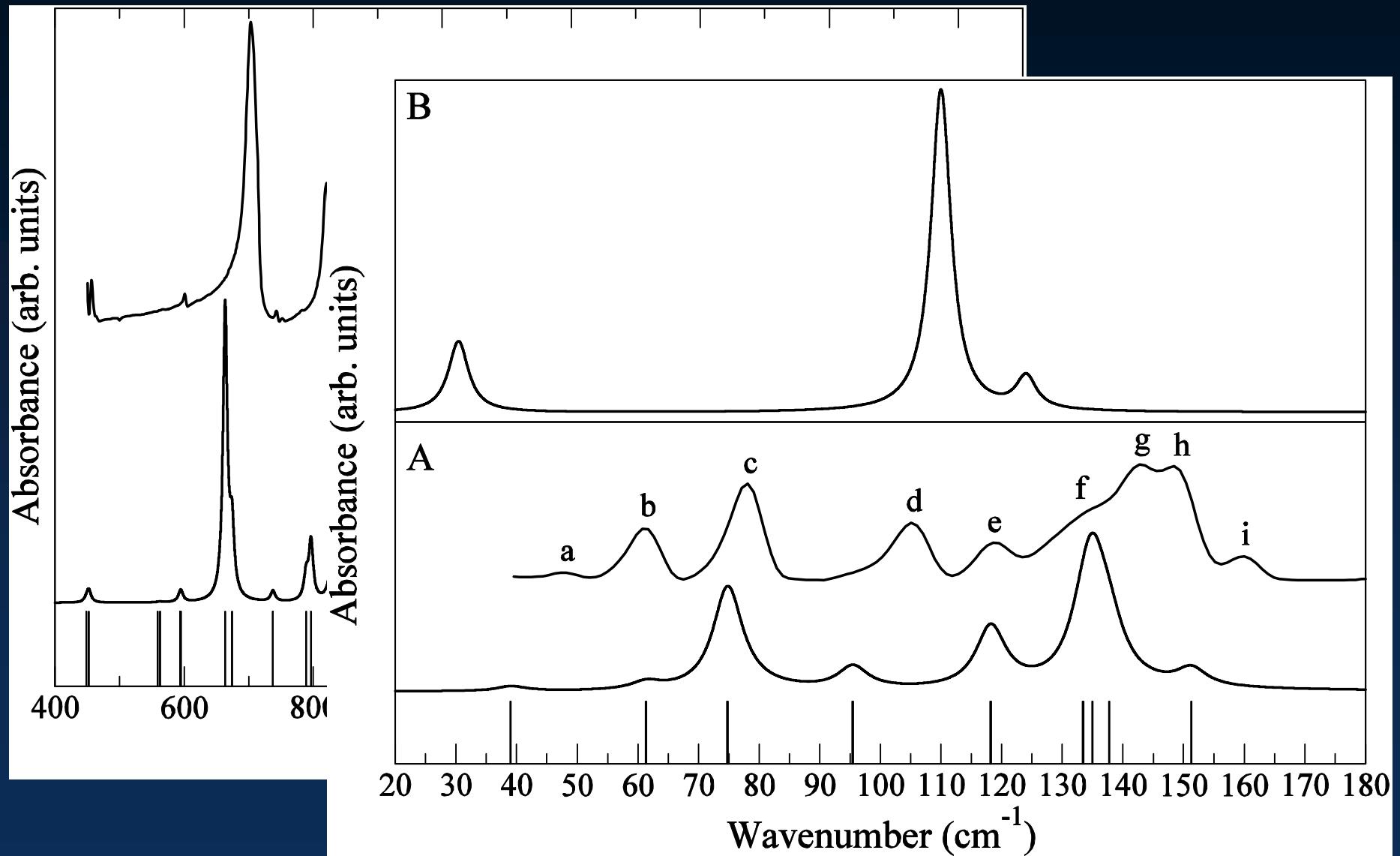
2T: Supercell size and anharmonicity-low frequency



2T: Supercell size and anharmonicity-high frequency



Bithiophene - INS vs IR



Benzoic acid

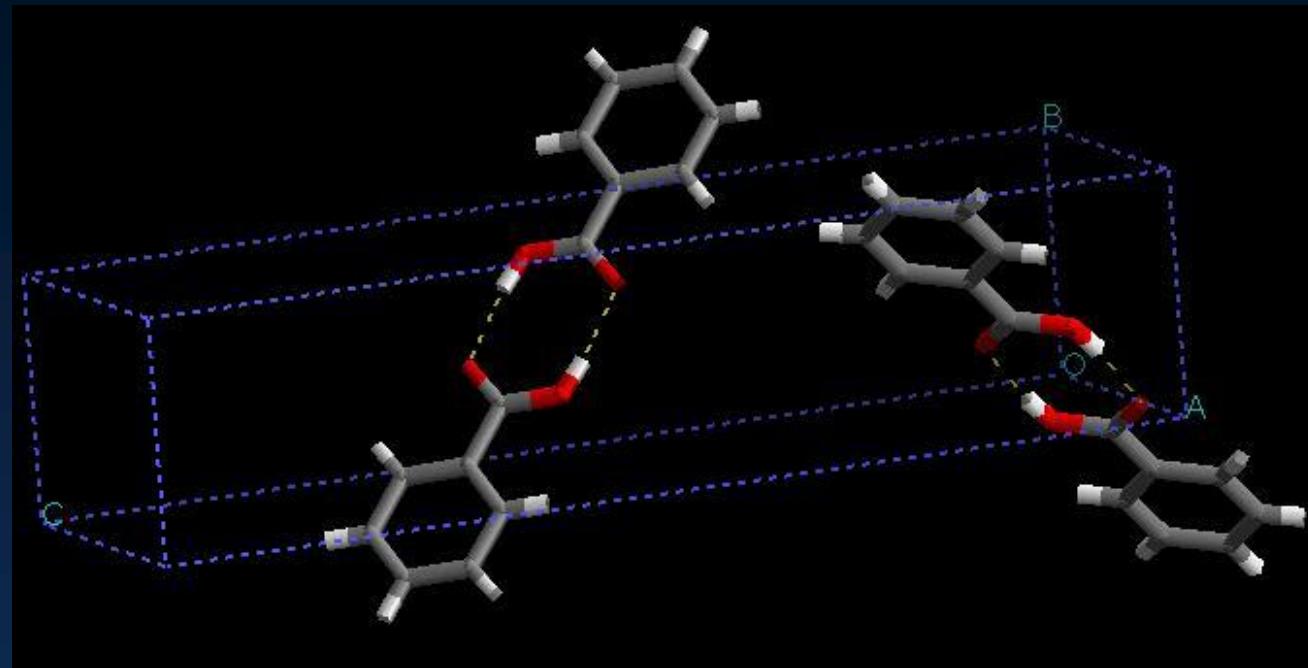
vibrations and proton tunnelling

- gamma point modes
- Gaussian vs. VASP
- a little bit of dispersion

(Marie Plazanet, Tony Horsewill, Peter Trommsdorff)

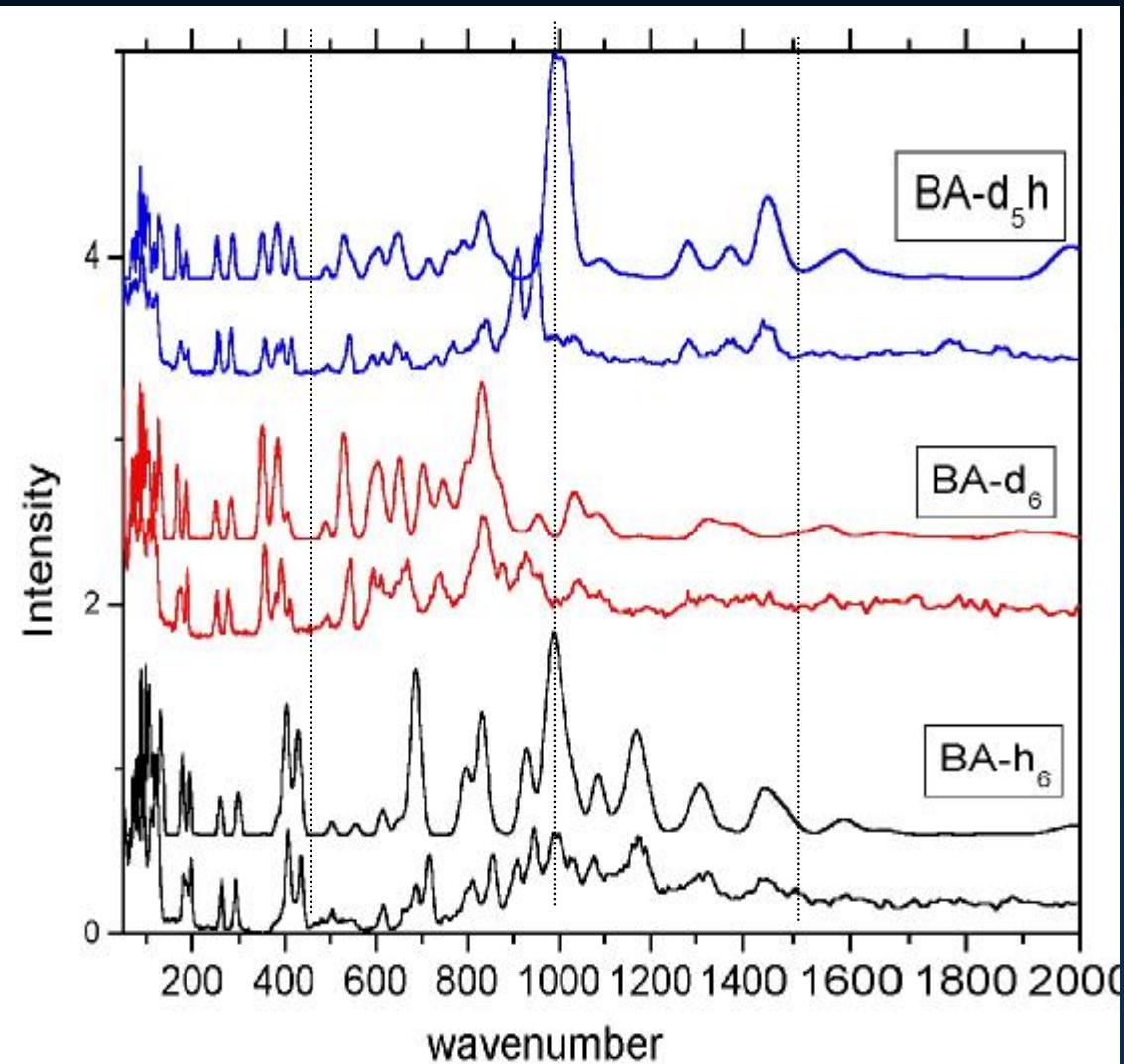
J.Chem.Phys 2001, Chem.Phys.Lett 2002

Crystal Structure of Benzoic Acid



O-H ...O distance = 2.61Å (Chick
Wilson et al., single-crystal, neutron
diffraction at ~10 K - J. Chem. Soc.
Faraday Transactions 92 (1996) 5051)

Vibrations in Benzoic Acid Dimers

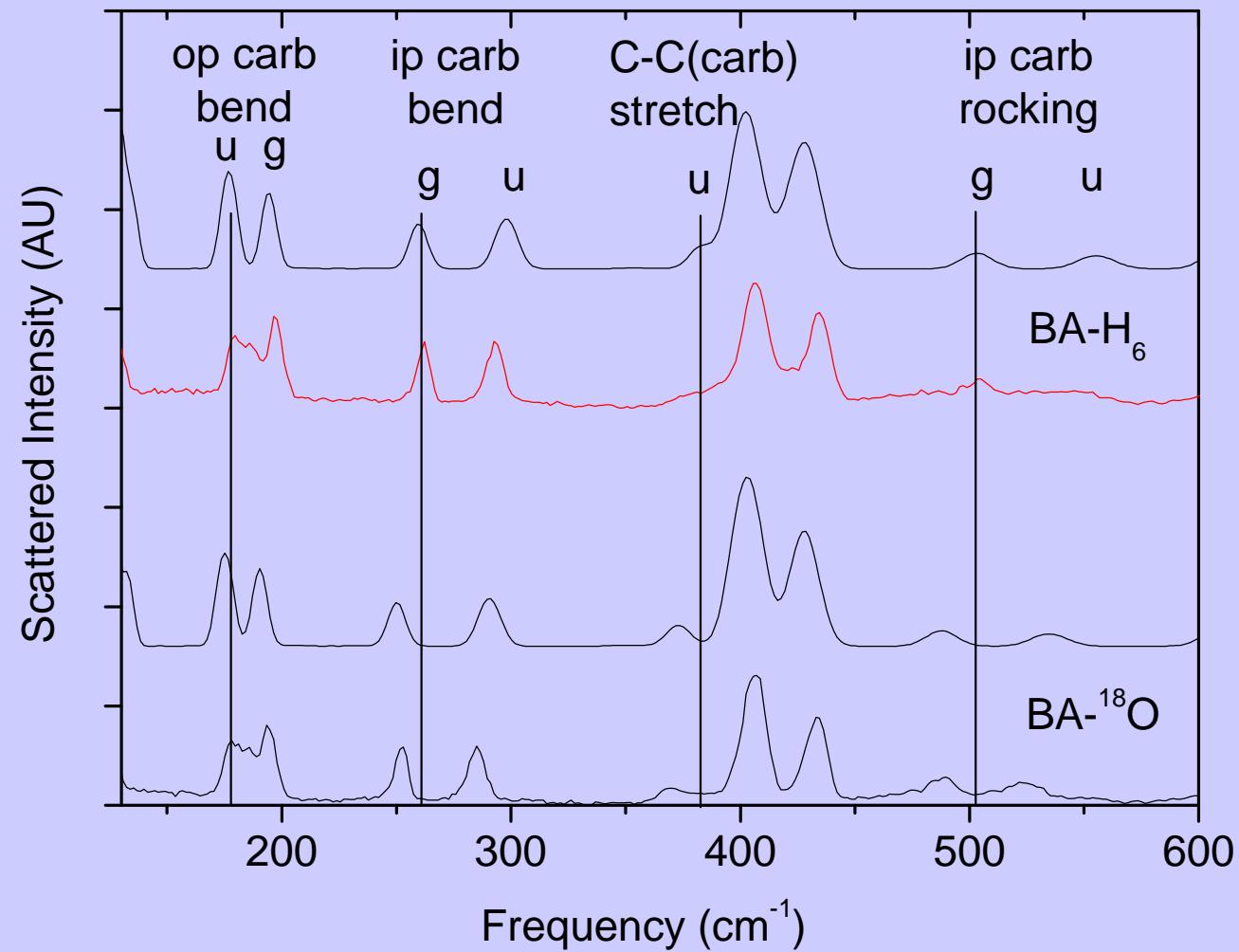


Dimer (Gaussian) gives excellent results, with only small overestimate of acid proton modes

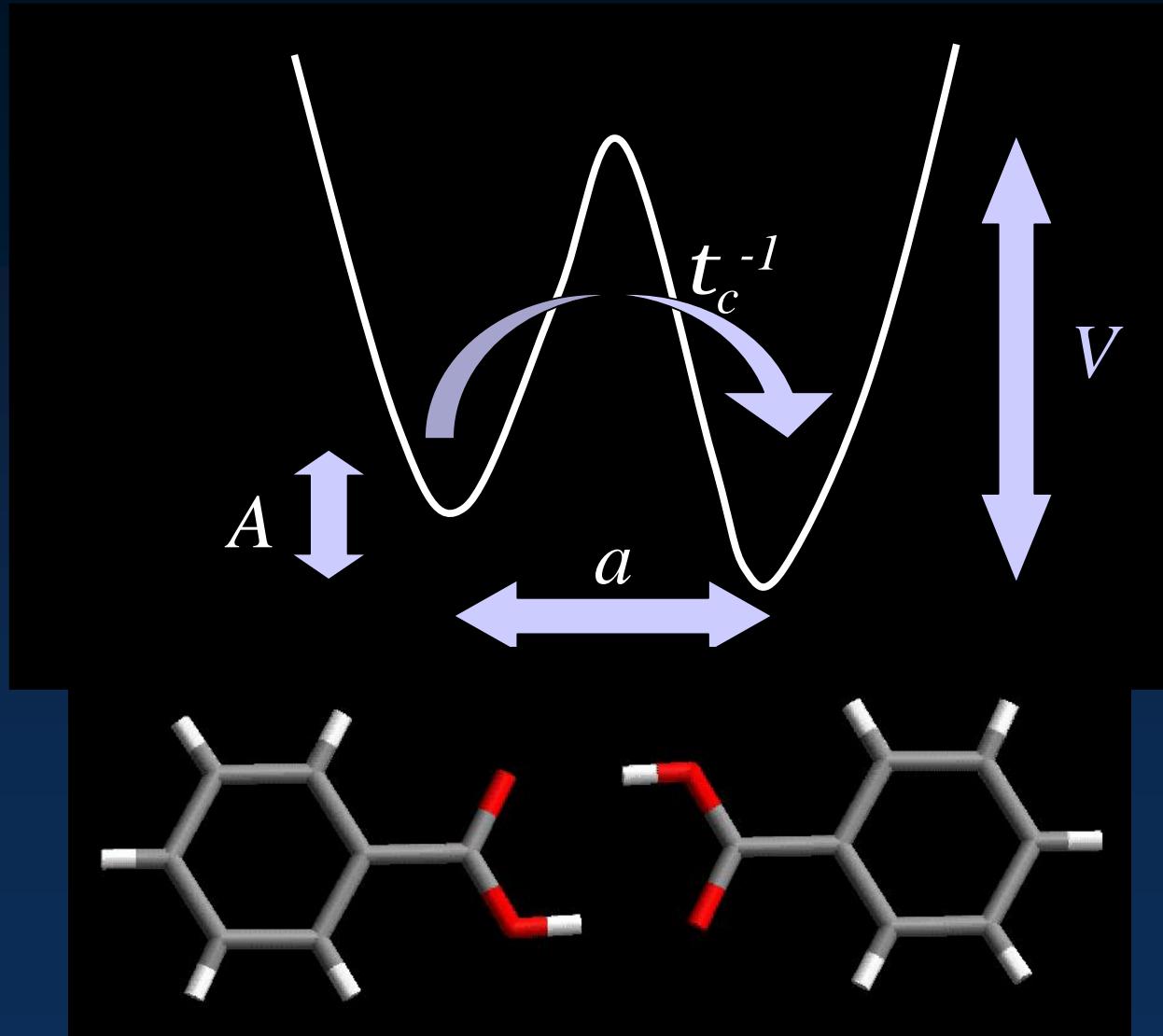
Periodic DFT (VASP) is as good if not better

Overestimated h-bond modes: O-H ... O = 2.55 Å & anharmonicities

How good is PW-DFT? ^{16}O -BA ® ^{18}O -BA



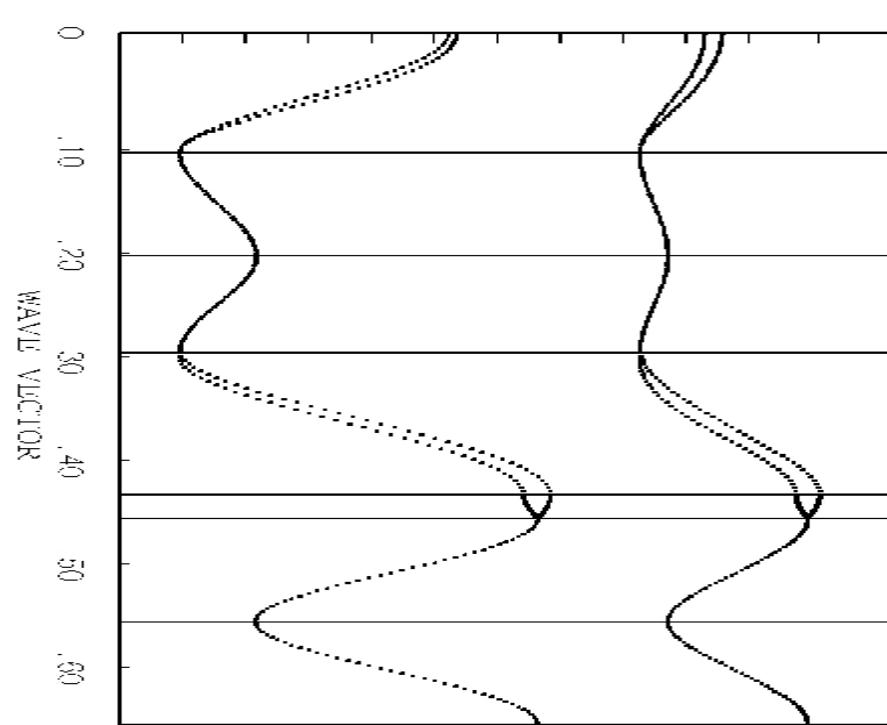
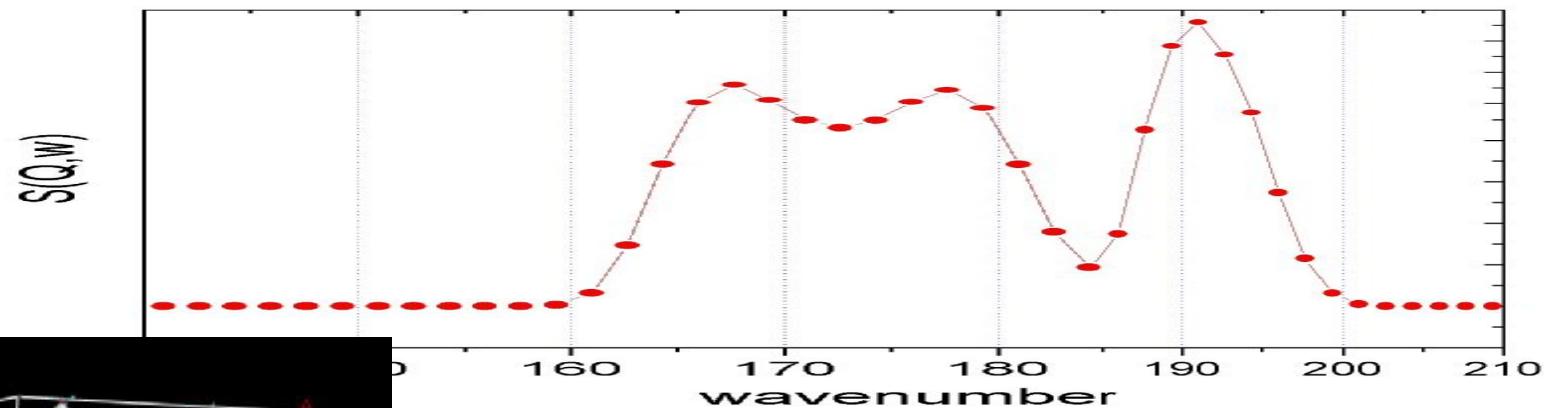
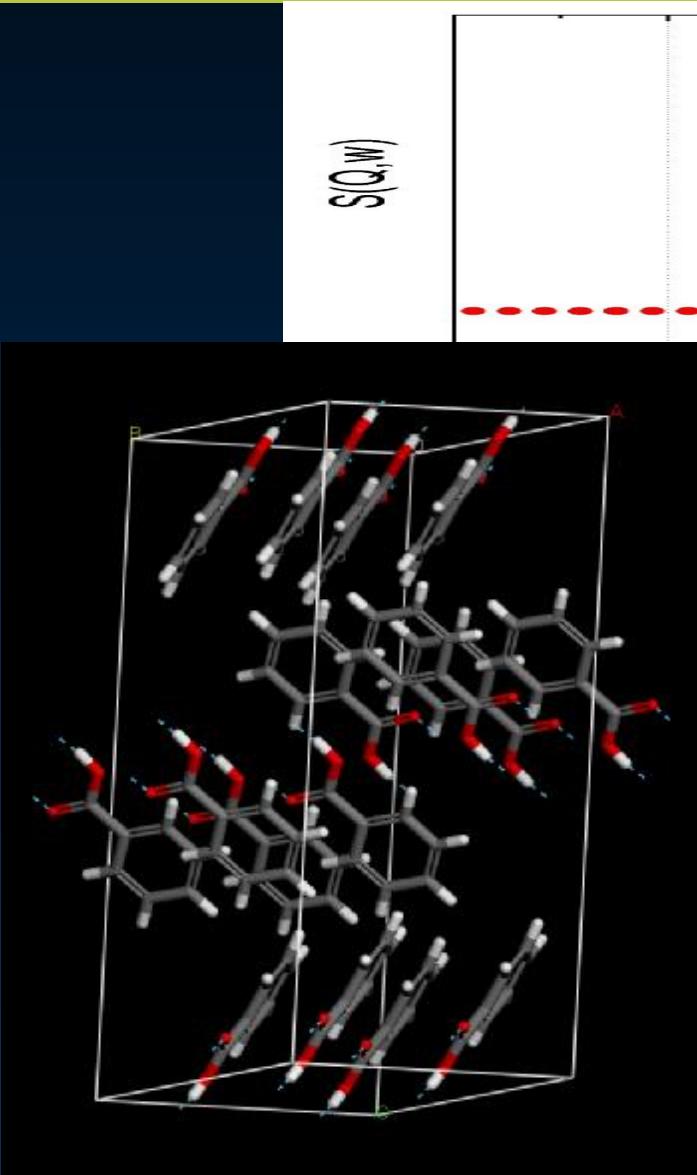
Vibrations and Proton Transfer in BA



Activation energy from NMR and QENS = 125 cm^{-1}

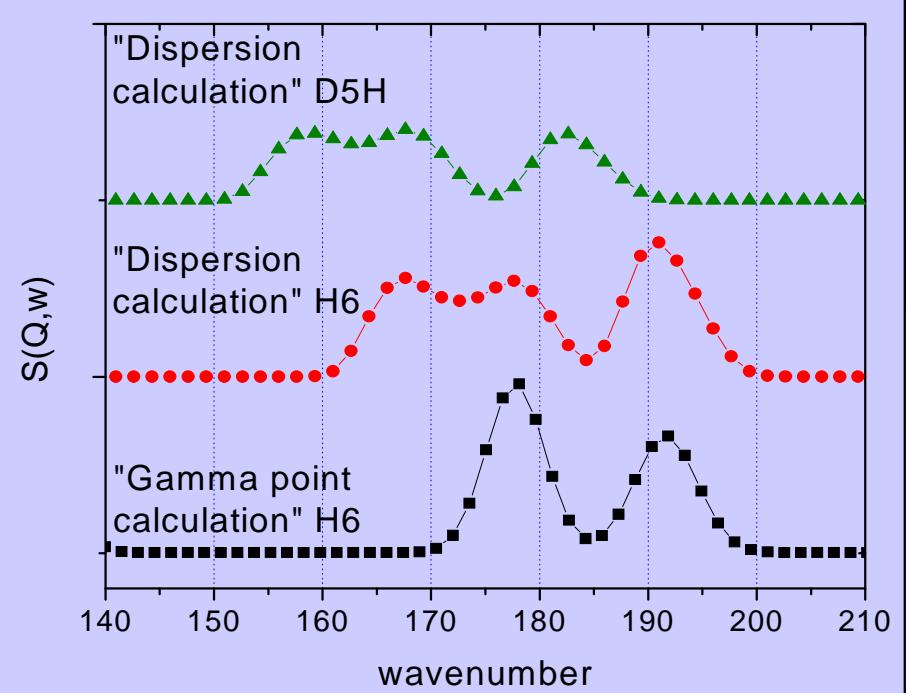
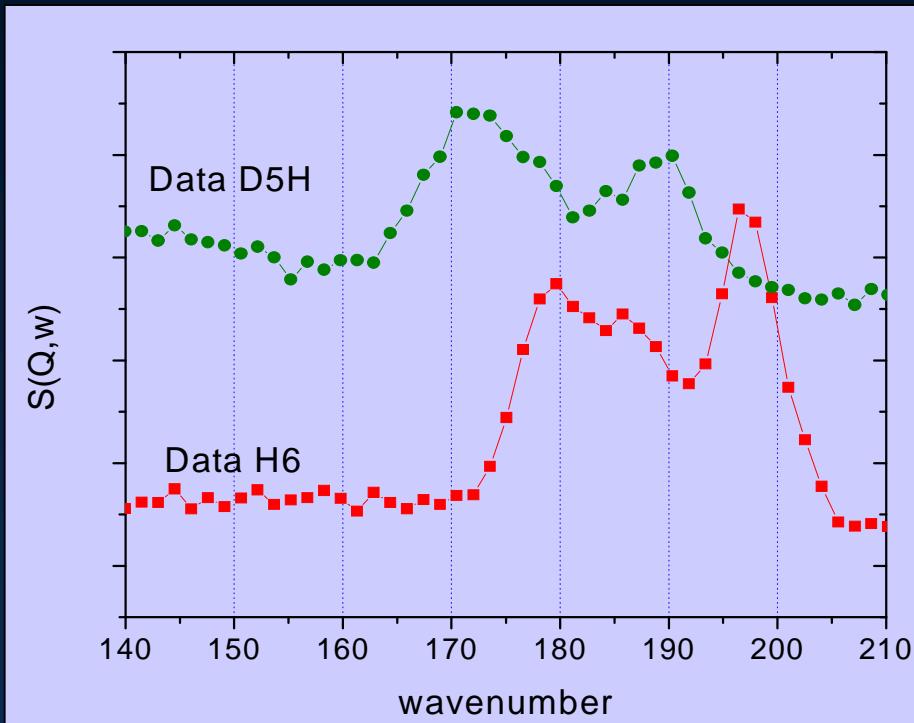
Active modes from VASP = 129 cm^{-1} & 136 cm^{-1}

... AND LATTICE DYNAMICS CALCULATIONS

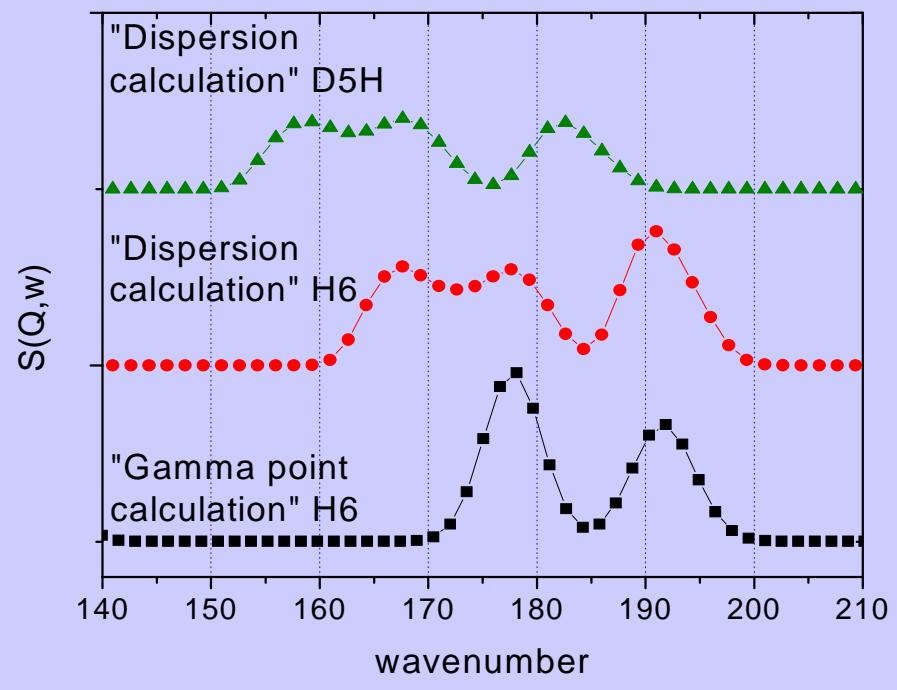


E A Y Z C C F

... AND LATTICE DYNAMICS CALCULATIONS

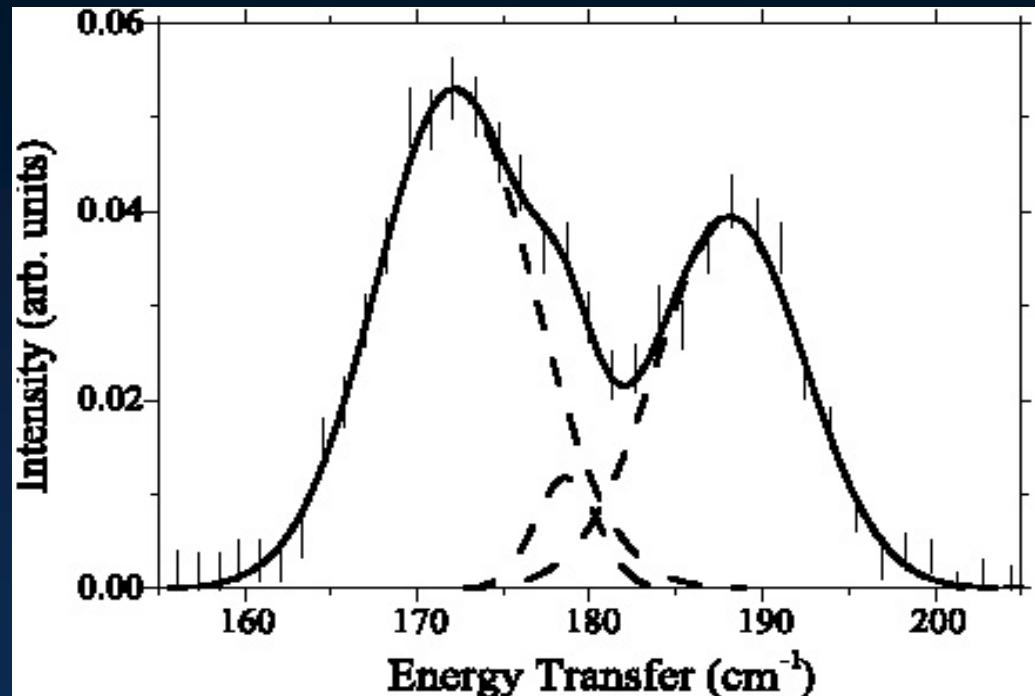


SO ... DISPERSION OR TUNNELLING AT 180cm⁻¹?



Tunnelling splitting $\sim 0.25\text{cm}^{-1}$, PRL 63 (1989) 1432

Dispersion at 180cm⁻¹, Chem. Phys. Lett. 364 (2002) 34



Tunnelling peak at 180cm⁻¹, F.Filloux, Chem. Phys., 276 (2002) 181

Urea

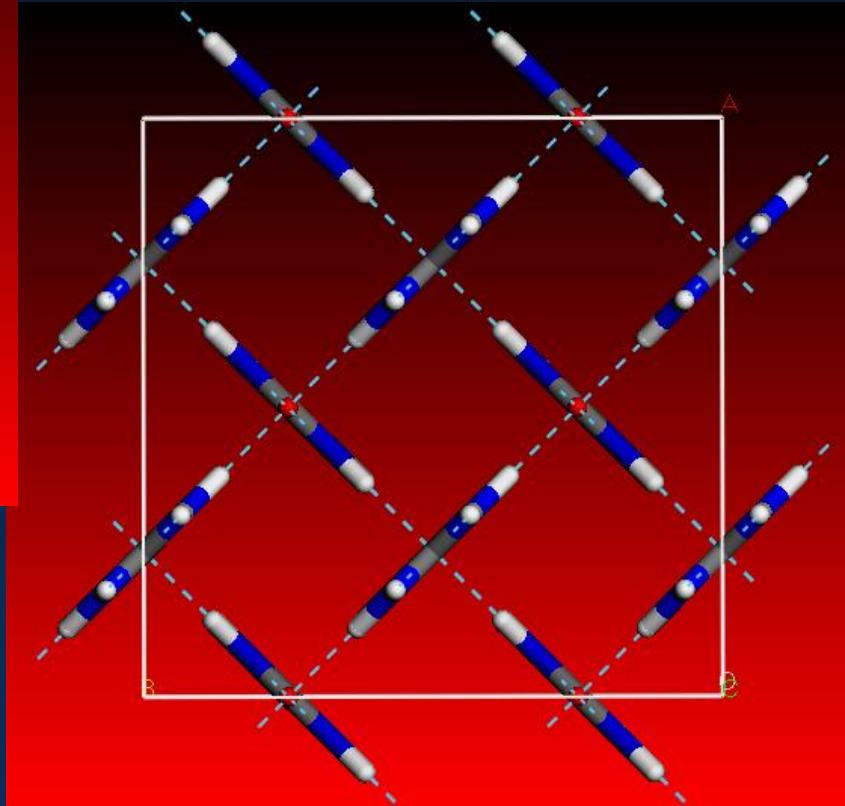
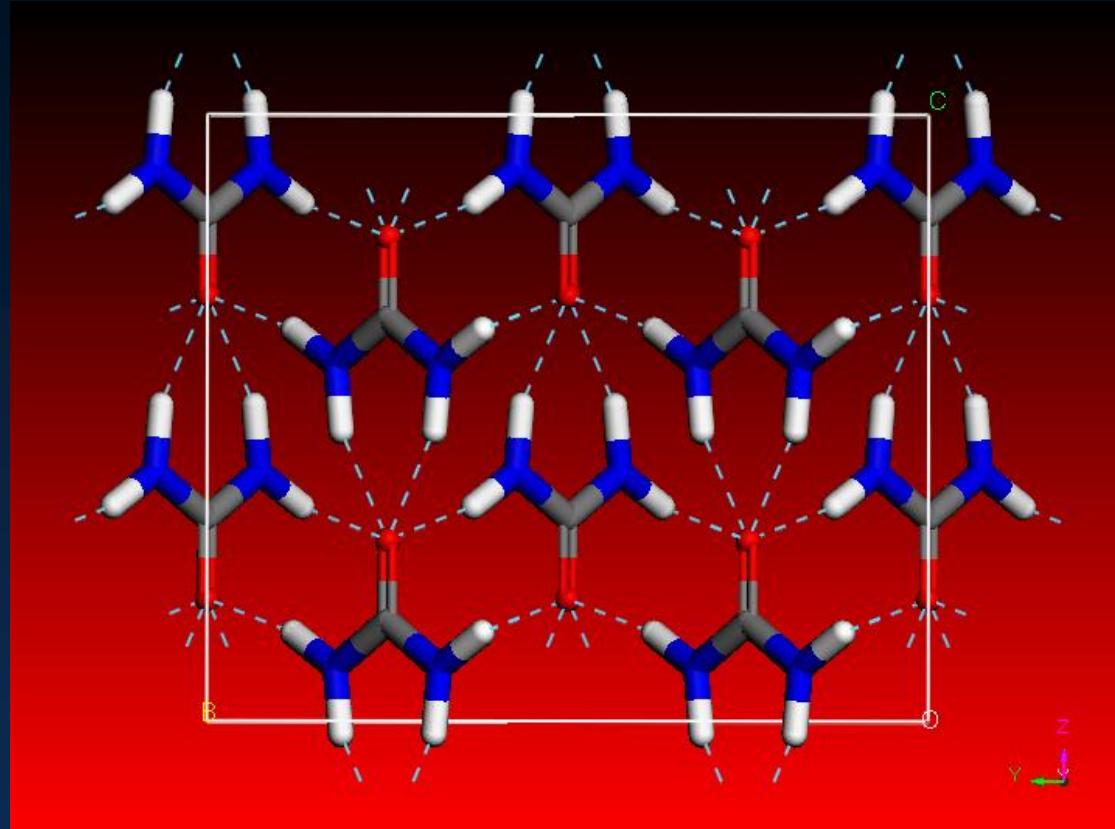
extended 3D hydrogen bond network

-multi-phonons

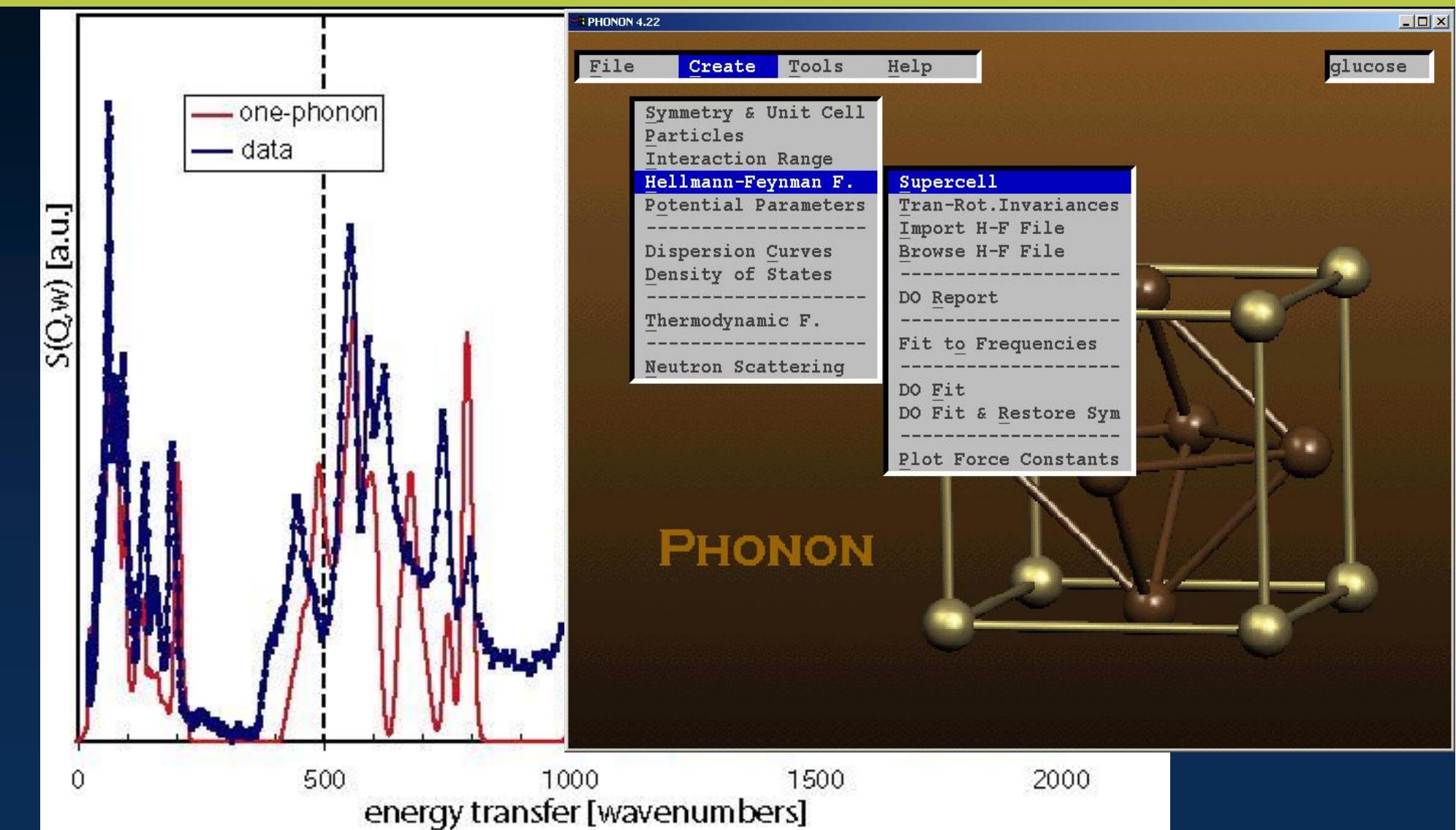
(Krzysztof Parlinski)

Chem.Phys 2002

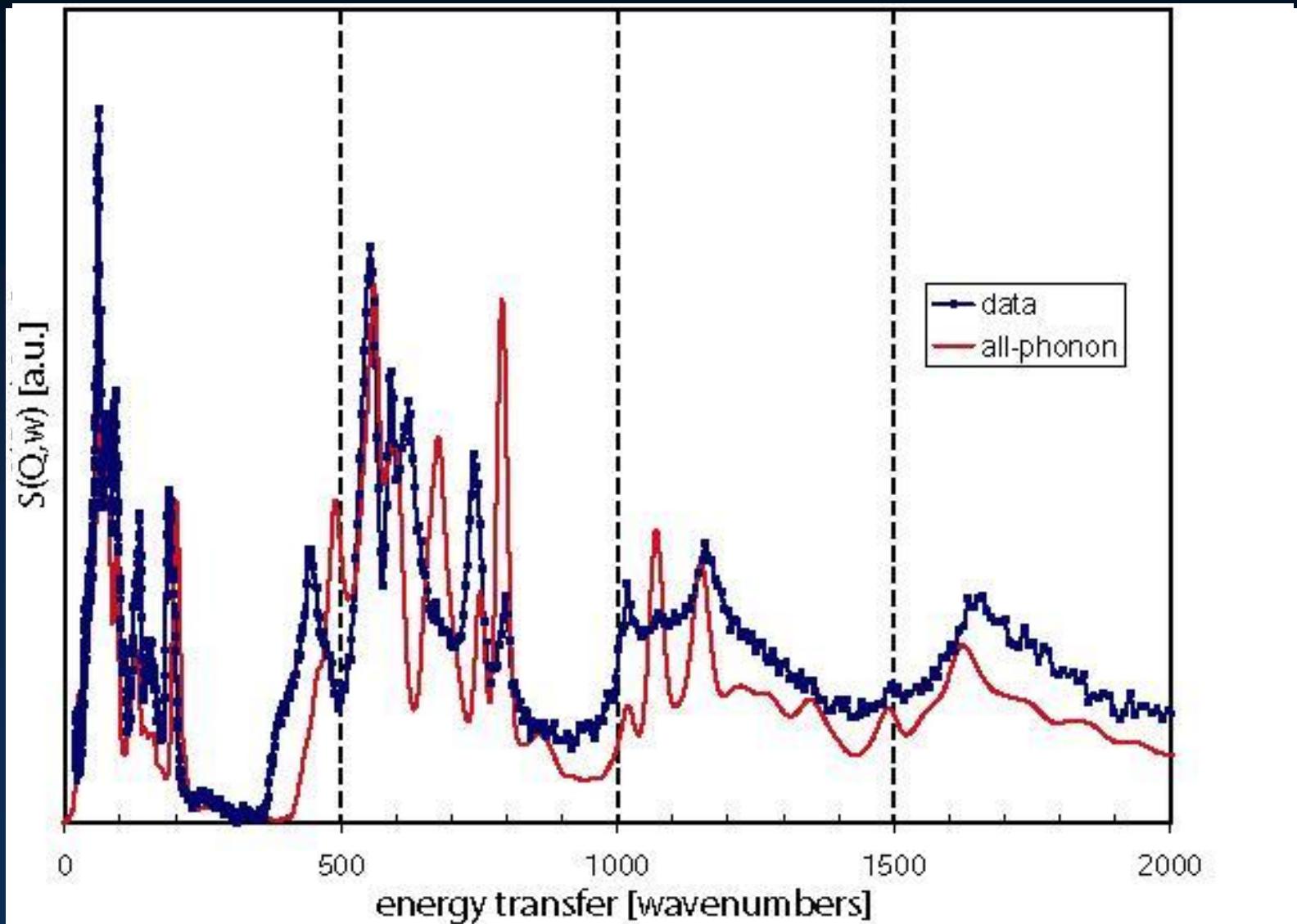
UREA - Dispersion in h-bond networks



UREA - Dispersion in h-bond networks



UREA - multi-phonon spectra



UREA - Dispersion in h-bond networks (400-800cm⁻¹)

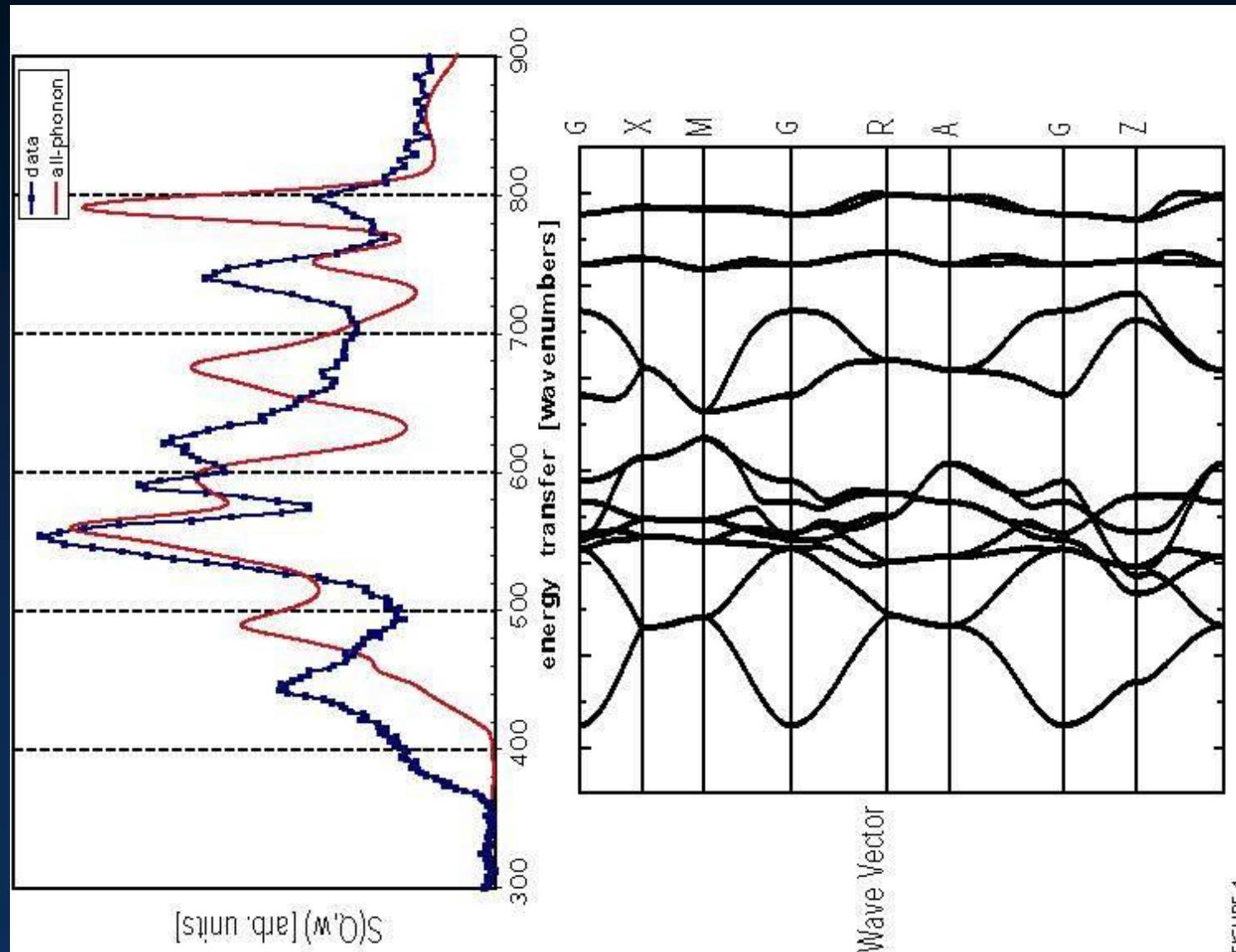
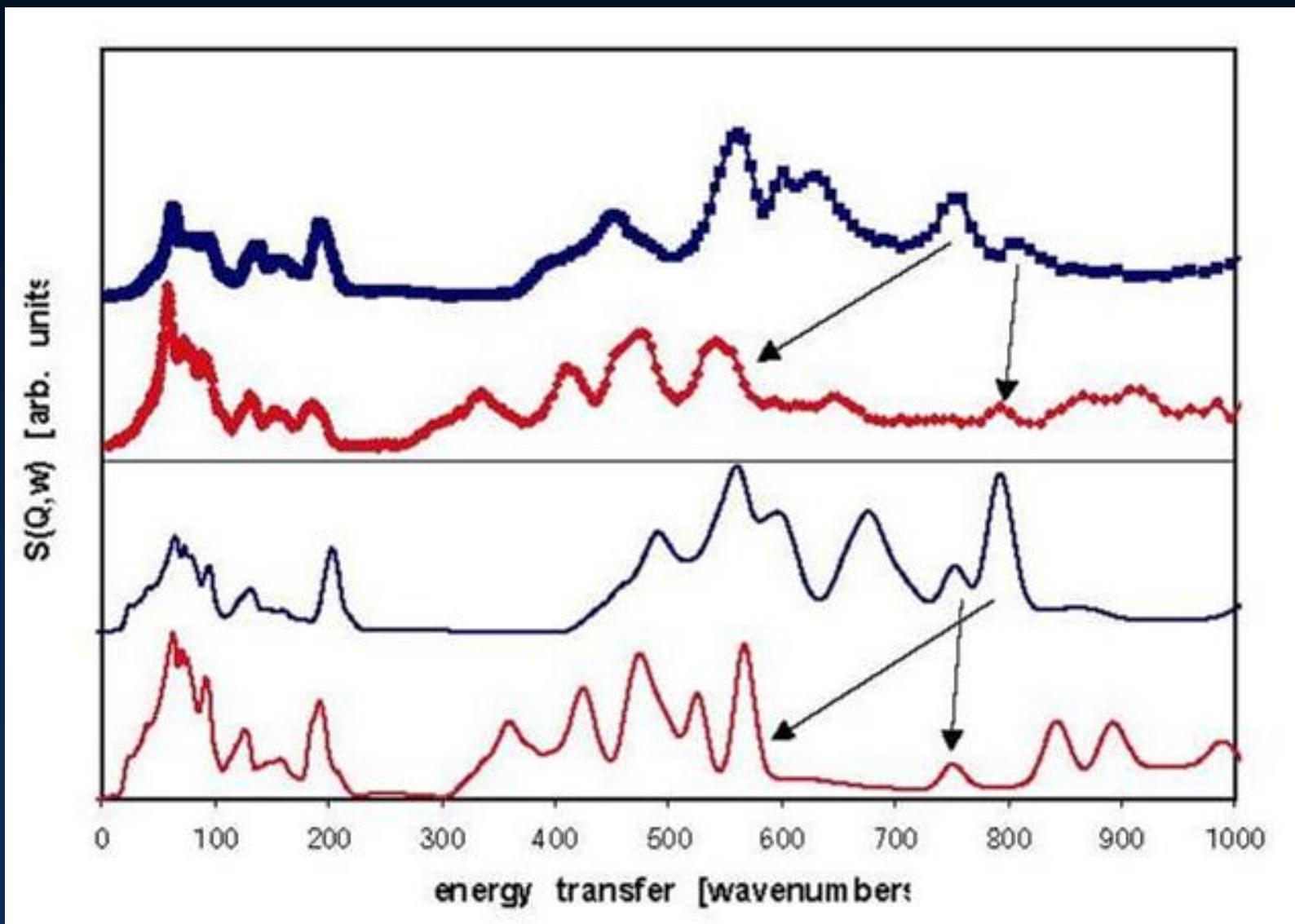


FIGURE 4

UREA - H/D substitution



Conclusion and Perspectives

Good insight from VASP + PHONON approach

Continue to do this wherever possible

For more complex systems ...

- 1/ more inequivalent atoms (PHONON)
- 2/ often only vDOS data is available
- 3/ ~1000's steps of MD are more efficient than series of force calculations
- 4/ average over many structures
- 5/ no harmonic approximation in determining $S(K,\omega)$
- 6/ better suited to liquid and amorphous systems

