

Lattice dynamics of ReO_3 from the quasiharmonic approximation

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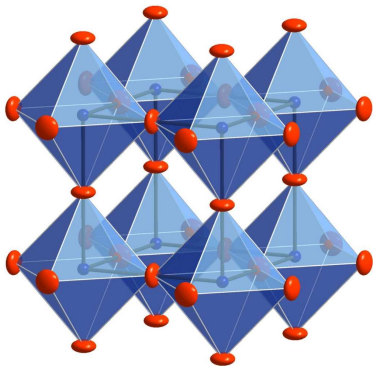
Origins of NTE

- **magnetostriction** - ferroelectric materials
- **valence transition** - intermetallic compounds, fulleride materials
- **low-frequency rigid unit modes (RUM)**
 - oxide-based framework materials
 - CN-based compounds
- materials with open framework structure - usually insulators or semiconductors

ReO₃

- metal - low electrical resistivity (comparable to Cu or Ag)
- metallic conductivity due to delocalized 5d-Re electrons
- highly covalent Re-O bond ⇒ **covalent metal**
- properties of the lattice - **not typical for metallic phase**

Network of ReO_3 octahedra



- space group $Pm\bar{3}m$
- lattice constant $a = 3.74 \text{ \AA}$

perovskite-type compound (ABO_3)
with A site vacant

Rigid rotation of octahedra

- anisotropy of thermal motion
- NTE
- structural phase transitions

Rigid unit modes and M-type phonon

Experimental observations

$Pm\bar{3}m \Rightarrow Im\bar{3}$

- 5.2 kbar (RT)
- 3.5 kbar (150 K)
- 2.5 kbar (4.2 K)

Intermedite phase/phases?

$Pm\bar{3}m \rightarrow P4/mbm \rightarrow Im\bar{3}$

↓

$P4/mbm \rightarrow I4/mmm \rightarrow Im\bar{3}$

NTE

- 100-340 K
laser interferometry
- below 100 K
x-ray diffraction, EXAFS
- up to 200 K
neutron diffraction

NTE depends on sample quality
(static disorder)

Experiment and calculations

Inelastic neutron scattering

- IN4C TOF spectrometer @ ILL
- $E_i = 16.9$ meV (2.2 Å)

Resolution

- ~ 0.6 meV @ elastic scattering
- ~ 4 meV @ 50 meV
- ~ 10 meV @ 100 meV

Corrections for multiphonon and DW contributions

MUPHOCOR by W.Reichardt
Karlsruhe Institute for Nuclear Physics

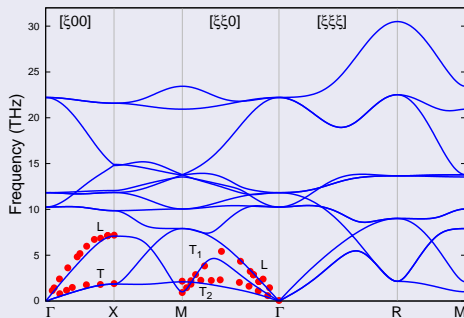
DFT calculations

- VASP code, PAW PP
- LDA approximation
 - $a_{\text{CAL}} = 3.7467$ Å
 - $a_{\text{EXP}} = 3.7477$ Å
- 2x2x2 s.c. (32 atoms)

Phonons

- Direct method - **PHONON**
- Thermal properties - **QHA**

Dispersion relations for ReO_3 at ambient pressure



Experiment: J. D. Axe *et al.* PRB 31, 663 (1985)

$T_1(\xi\xi 0)$ mode undergoes softening near the M-point

$$M \equiv T_1\left(\frac{1}{2}, \frac{1}{2}, 0\right)$$

Cal.	150 K	300 K
1.007	0.822	0.933

$$R \equiv T\left(\frac{1}{2}, \frac{1}{2}, \frac{1}{2}\right)$$

Calculation: 2.146 THz

Γ -point modes

- **IR active**

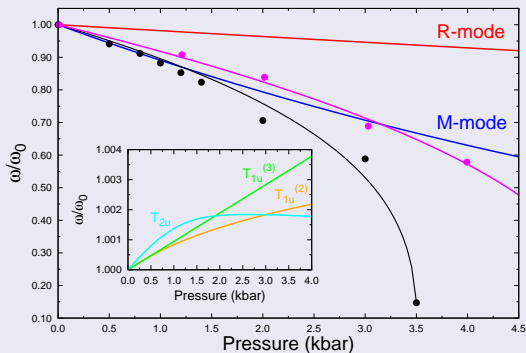
$$T_{1u}^{(2)} \rightarrow 10.255 \text{ THz}$$

$$T_{1u}^{(3)} \rightarrow 22.217 \text{ THz}$$

- **silent**

$$T_{2u} \rightarrow 11.825 \text{ THz}$$

Rotational modes condense under compression



Grüneisen constants

$T_{1u}^{(2)}$	$T_{1u}^{(3)}$	T_{2u}
1.24	2.11	1.03

Grüneisen constants

Mode	Cal.	Exp.
M	-291	-242 ^[1] -180 ^[2]
R	-46	-

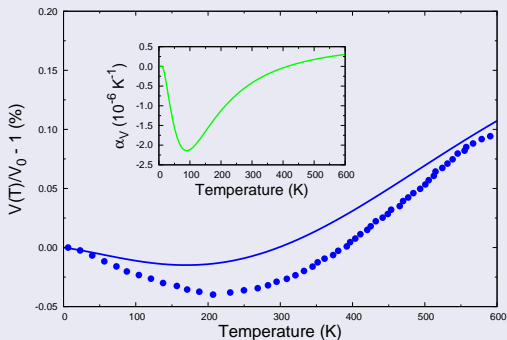
Softening rates (THz/kbar)

M	R
-0.12	-0.04

[1] T. Chatterji *et al.* PRB 79, 184302 (2009)

[2] J. D. Axe *et al.* PRB 31, 663 (1985)

Thermal expansion of ReO_3



[1] T. Chatterji *et al.* PRB 78, 134105 (2008)

[2] T. Chatterji *et al.* PRL 94, 241902 (2009)

NTE

Cal.	Neutron	LDC
170 K	200 K	350 K

Average α_V below NTE-PTE crossover temperature

$$\langle \alpha_V \rangle = -1.2 \times 10^{-6} \text{ 1/K}$$

QHA vs LDC

- similar expansivity below NTE-PTE
- PTE regime $\alpha_V(\text{LDC}) \sim 2 \alpha_V(\text{QHA})$

NTE of ReO_3 and atomic thermal motion

NTE and static disorder in ReO_3

- large contribution from static disorder \Rightarrow **diminished NTE**
- samples with less static disorder
 - high NTE-PTE crossover temperature (294 K)
 - quite large negative thermal expansivity (-1.0×10^{-6} 1/K)

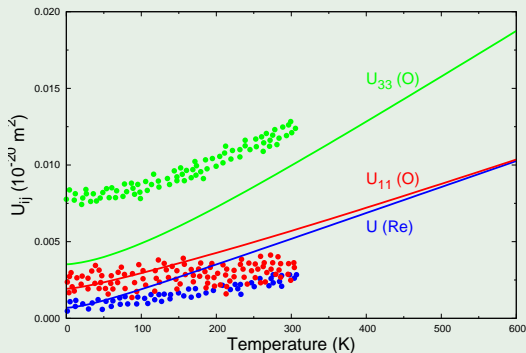
Mean-squared vibrations of ReO_3

- Experimental U_{ij} contain static and dynamic contributions
- Present calculations - dynamic component
- Re ($\bar{3}m$) \Rightarrow **isotropic** thermal motion $\Rightarrow U(\text{Re})$
- O ($4/mmm$) \Rightarrow **anisotropic** thermal motion

$U_{11}(\text{O})$	oxygen vibrations \parallel to Re-O bond
$U_{22}(\text{O}) = U_{33}(\text{O})$	oxygen vibrations \perp to Re-O bond

Atomic thermal motion in ReO_3

Thermal vibrations of Re and O



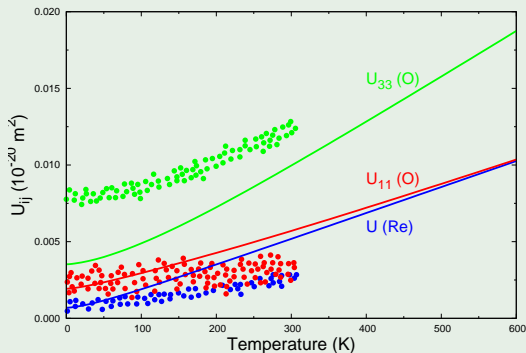
T. Chatterji *et al.* PRB 78, 134105 (2008)

Questions

- 1 $U_{11}(\text{O}) \approx U(\text{Re})$?
 $\Phi(\text{Re}) \sim 1.3 \times \Phi_{xx}(\text{O})$
- 2 $U_{33}(\text{exp.}) \gg U_{33}(\text{cal.})$
 LDC fails either !
 STATIC DISORDER

Atomic thermal motion in ReO_3

Thermal vibrations of Re and O



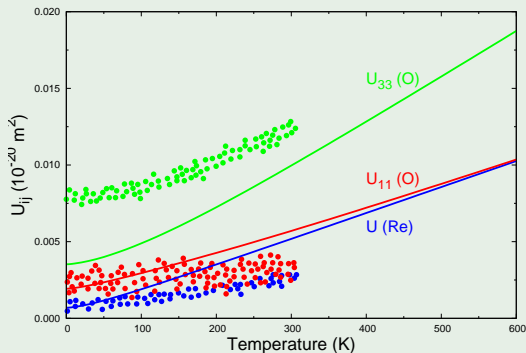
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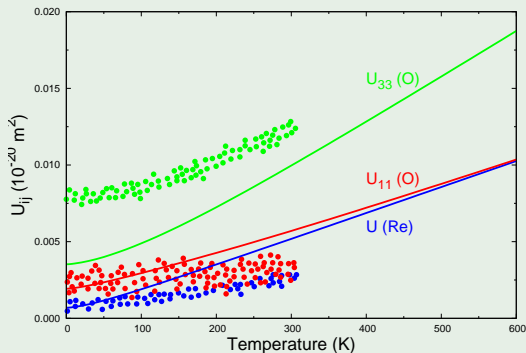
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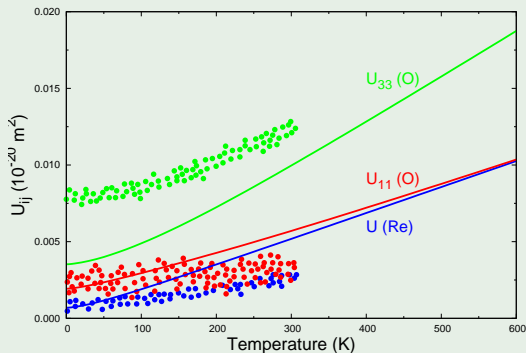
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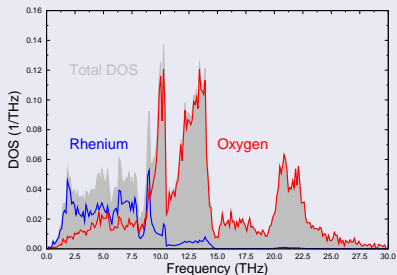
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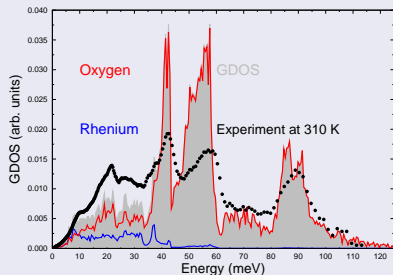
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STATIC DISORDER

Thermal motion and static disorder of ReO_3

DOS



GDOS



$\alpha(\text{Re})=0.062$ b/amu, $\alpha(\text{O})=0.265$ b/amu
scattering dominated by oxygen (93%)

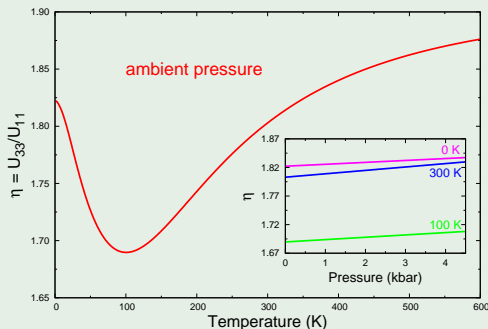
- static disorder breaks propagation of phonons throughout lattice
- shift of intensity for peaks at 41 meV and 55 meV into continuous distribution at low energies (10-35 meV)

Anisotropy of the oxygen thermal vibrations

- high quality samples
 - lower gap between U_{33} and U_{11}
 - diminished anisotropy $\eta = U_{33}/U_{11}$
 - low temperatures: η (exp.) \approx η (cal.)
-
- η related with anisotropy in the force constants at O site
 - $\Phi_{zz} \approx \frac{1}{2} \Phi_{xx}$
↓
 - larger amplitude of thermal vibrations in direction perpendicular to Re-O bond than along the bond
-
- highly anisotropic motion in a direction consistent with M-phonon mode
↓
 - coordinated rotations of ReO_3 octahedra

Anisotropy vs temperature and pressure

Anisotropy vs temperature



$T = 100 \text{ K}$

- anisotropy minimum
 - expansivity minimum
-
- NTE coincides with more isotropic thermal motion of O atoms
 \Downarrow
 - reduction/enhancement in η - additional driving force converting ReO_3 from NTE to PTE

anisotropy vs pressure

- simultaneous hardening of $\Phi_{xx}(\text{O})$ and softening of $\Phi_{zz}(\text{O})$
- reduction of U_{11} and enhancement of $U_{33} \Rightarrow$ increase of anisotropy

Summary and conclusions

- 1 Large negative values of Grüneisen constants for M and R phonons
- 2 M-phonon softens more rapidly than R-phonon with crystal compression
- 3 Anisotropy of the oxygen thermal motion in ReO_3
 - follows from the anisotropy of the force constants at the oxygen site
 - consistent with the M-phonon
 - contributes to NTE of ReO_3
 - NTE is enhanced when oxygens perform more isotropic thermal motion
- 4 Anisotropy - additional force driving the low-pressure phase transitions
- 5 Details can be found in PRB 82, 104301 (2010)